



## Honokiol as an Effective Antimicrobial Compound against Causative Agent of American foulbrood, *Paenibacillus larvae*

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### Abstract

Recently, number of honeybees (*Apis mellifera*) has visibly decreased because they are vulnerable to some diseases like American foulbrood disease. American foulbrood disease, which is caused by *Paenibacillus larvae*, is emerged as great cause of decrease in number of honeybees. After antibiotic-resistant strain emerged, it is now more difficult to treat those pathogens successfully. Researches on finding alternative antibacterial compound are ongoing. In this study, we examined the antibacterial effect of honokiol on *P. larvae*. Honokiol showed great antibacterial effect with minimum inhibitory concentration of 12.5 µg/mL and minimum bactericidal concentration of 50 µg/mL. An agar diffusion test also confirmed the anti-*Paenibacillus larvae* activity of honokiol with an inhibitory zone of 9±0.5 mm. Since honokiol is known to interact membrane of some bacteria, we measured 260 nm absorbing particles, which could be induced by leakage of cells, and confirmed that the leakage of *P. larvae* occurred in dose-dependent manners. However, result of crystal violet assay suggested that honokiol has only mild anti-biofilm formation effect on *P. larvae*, which means honokiol controls the bacteria by inducing the bursting of membrane. Finally, an additive effect of honokiol with tetracycline and terramycin was found using a checkerboard assay with a fractional inhibitory concentration index value of 0.5.

### Keywords

*Paenibacillus larvae*, Honokiol, Antibacterial agent, Membrane leakage

## INTRODUCTION

Honeybees (*Apis mellifera*) are indispensable in our ecosystem and some important industries. Their role as pollinators help many plants to prosper which made them value over \$14.6 billion in crop and fruit industry (Morse and Calderone, 2000). However, due to diseases like American foulbrood (AFB) the population of honeybees has been steadily decreasing and caused many problems.

*Paenibacillus larvae* is a causative agent of AFB (White, 1906). Since it can be easily spread over entire colony and could effect on larva of honeybees, it has been a great problem for honeybees. AFB has now been spread all over the world and appeared frequently (Ge-

nersch *et al.*, 2006). Even though terramycin has been used to treat *P. larvae* (Gochnauer, 1951), unfortunately, *P. larvae* resistant to terramycin has arose over the world (Miyagi *et al.*, 2000).

So far, many studies have been conducted to reduce the population of *Paenibacillus larvae*. However, it had many difficulties because *P. larvae* is resistant to heat and various chemicals. Therefore, it is urgent to find new antimicrobial substances, which can kill them effectively, rather than stopping the growth of *P. larvae*.

*Magnolia obovata*, which is a tree used as stomachic herb in Korea, Japan and China (Huang, 1998). Honokiol is a compound, which is known to have a great antifungal activity and can be easily found in the bark

of *M. obovata* (Bang *et al.*, 2000). It is also known as great anxiolytic agent (Maruyama *et al.*, 1998).

The present study provided an antibacterial activity of honokiol and proof that it could serve as an effective inducer of bursting of *P. larvae*. This study also showed an additive effect of honokiol with well-known antibiotics, terramycin and tetracycline and we strongly suggested that honokiol could be used as a novel chemical with anti-*P. larvae* activity.

## MATERIALS AND METHODS

### Microorganism strain and growth conditions

*P. larvae* (strain ATCC 9545) were developed in Mueller-Hinton broth, yeast extract, potassium phosphate, glucose and sodium pyruvate (MYPGP broth) at 37°C with shaking overnight (Dingman and Stahly, 1983). Before use, cells were centrifuged at 3000 rpm for 10 minutes. After old media was replaced with new media, cells were incubated again for 2 hours before use.

### Compound preparation

All compounds used in this study, honokiol (CAS: 35354-74-6, Chemfaces), terramycin (CAS: 2058-46-0, Tokyo Chemical Industry) and tetracycline (CAS: 60-54-8, Sigma-Aldrich), were dissolved in dimethyl sulfoxide (DMSO) to 10 mg/mL before use.

### Investigation of antimicrobial effects of compounds

Minimum inhibitory concentration (MIC) and Minimum bactericidal concentration (MBC) were determined prior to other experiments. To determine MIC, a two-fold micro-dilution method was used as described in previous studies (Zgoda and Porter, 2001).  $10^5$  CFU/mL of *P. larvae* were added into each well of a 96 well plate and the compound in each lane were then diluted two-fold from 100 µg/mL to 0.156 µg/mL. After a 48-hour incubation at 37°C, the minimum inhibitory concentration of each compound was measured.

To determine MBC, broth from clear wells after MIC test were spread on MYPGP agar plate. Plates were

placed in 37°C for another 48 hours and Minimum bactericidal concentration of each compound was measured.

### Disk diffusion test

Antibacterial sensitivity of compounds was determined using the agar diffusion method (Fiebelkorn *et al.*, 2003). *P. larvae* was diluted to an optical density of 0.1 at 600 nm and they were spread onto an MYPGP agar plate. 50 µg of each compound was added to a 0.3 mm-radius paper disk and placed on the MTPGP agar plate. Cells were incubated for 24 hours at 37°C and diameter of the clear zone was measured.

### Checking leakage with 260 nm absorbing particles

In order to verify the leakage of cells after treatment of the compound, 260 nm absorbing particles were measured. To stop further proliferation, cells were incubated in PBS for 2 hours. Compounds were then treated with concentration ranging from 3.12 to 50 µg/mL for 1 hour. Lysozyme as a positive control was treated to determine maximum optical density after bursting of the cells, each concentration of the compound was measured without cells. Lysate released after leakage was measured by microplate spectrophotometer (Epoch, USA) at an optical density of 260 nm. All values were revised by deducting the value of corresponding negative control (Chen and Cooper, 2002; Junaidah *et al.*, 2014).

### Visualizing the bursting with transmission electron microscopy

Transmission electron microscopy (TEM) visualized bursting *P. larvae* cells after treatment of honokiol on bacterial cells. The assay was performed using an 80 kV transmission electron microscope (JEM1010, JEOL) at NICEM (Seoul national university, South Korea). One hour before TEM imaging, honokiol treatment was carried out at a concentration of 6.25 µg/mL. Samples were loaded onto a grid and dyed with phosphotungstic acid (PTA). The grid was washed to remove any unattached cells. Cells that had not been treated with honokiol were used as a control.

### Crystal violet assay

A microtiter dish biofilm formation assay (O'Toole, 2011) was performed to determine inhibition of biofilm formation of *P. larvae* by honokiol. 100  $\mu\text{L}$  of  $10^5$  CFU/mL of *P. larvae* was added to each well of a 96 well-plate. Honokiol was treated with the concentration from 0.047 to 6.25  $\mu\text{g/mL}$ , the plate was placed at 37°C for 24 hours. After incubation, the plate was rinsed with distilled water and then stained with 0.1% crystal violet. Leftover crystal violet solution was removed after 20 minutes. Elution was done with 30% acetic acid and optical density of eluted solution was measured in new plate at 550 nm.

### Checkerboard assay

A checkerboard assay was performed to check for synergy effects between honokiol and two well-known antibiotics, tetracycline and terramycin (Schwalbe, 2007). All cells were treated with honokiol with tetracycline or terramycin with different concentration combinations. Honokiol was diluted horizontally as two-fold, while antibiotics were treated vertically by two-fold dilution. FIC index values were measured. The combination is considered synergistic when the FIC value is under 0.5, indifferent when the FIC value is in range between 0.5 to 2, and antagonistic when the FIC value is over 2.

## RESULTS

### Investigation of antimicrobial effect of compounds

Honokiol had great MIC value (12.5  $\mu\text{g/mL}$ ) compared to those of terramycin (25  $\mu\text{g/mL}$ ) and tetracycline (25  $\mu\text{g/mL}$ ) (Table 1). However, MBC value did not show much difference among the compounds (Table 2). They all had same MBC value of 50  $\mu\text{g/mL}$ .

### Disk diffusion test

Honokiol showed mild antibacterial activity against *P. larvae* (Table 3 and Fig. 1) with inhibitory zones of  $9 \pm 0.5$  diameters at 24 hours and 36 hours. Although it did not show better activity than terramycin ( $17 \pm 1$

**Table 1.** Minimum inhibitory concentration (MIC) values ( $\mu\text{g/mL}$ ) of honokiol, terramycin and tetracycline against *P. larvae* after 48 h treatment

Compound	MIC value ( $\mu\text{g/mL}$ )
Honokiol	12.5
Terramycin	25
Tetracycline	25

**Table 2.** Minimum bactericidal concentration (MBC) values ( $\mu\text{g/mL}$ ) of honokiol, terramycin and tetracycline against *P. larvae*

Compound	MBC value ( $\mu\text{g/mL}$ )
Honokiol	50
Terramycin	50
Tetracycline	50

**Table 3.** Antimicrobial susceptibility test results based on agar diffusion analysis shown with diameter of the cleared area

Chemical compound	Diameter (mm)	
	24 h	36 h
Honokiol	$9 \pm 0.5$	$9 \pm 0.5$
Tetracycline	$16 \pm 1$	$15 \pm 1$
Terramycin	$17 \pm 1$	$16 \pm 0.5$

at 24 hours and  $16 \pm 0.5$  at 36 hours) and tetracycline ( $16 \pm 1$  at 24 hours and  $15 \pm 1$  at 36 hours), it is definitely considerable as an antibacterial agent since it had great MIC value.

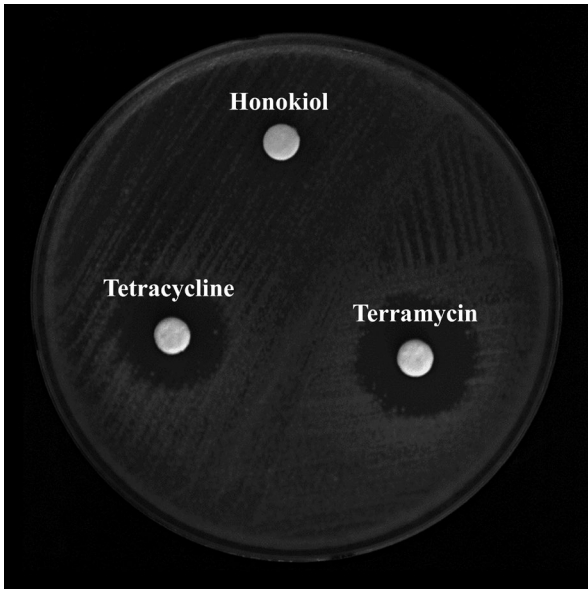
### Checking leakage with 260 nm absorbing particles

The release of 260 nm absorbing materials increased after treatment of honokiol (Fig. 2). When concentration of honokiol was lower than MIC value (12.5  $\mu\text{g/mL}$ ), it only had some subtle changes. However, at the concentration of 12.5  $\mu\text{g/mL}$ , 260 nm absorbing particles significantly leaked out from membrane ( $0.084 \pm 0.005$ ) and at the concentration of 50  $\mu\text{g/mL}$ , the optical density of 260 nm absorbing particles reached the same as optical density of lysozyme-treated samples ( $0.175 \pm 0.022$  and 0.168 respectively).

### Visualizing the bursting with transmission electron microscopy

Although there clearly was a leakage of *P. larvae* after honokiol treatment, it was not certain whether it in-

duced the bursting of cells or it made only some crack on the membrane. When it was visualized by TEM, honokiol surely induced a bursting of the cells (Fig. 3). Unlike control cells, honokiol-treated cells had their membrane destroyed and had their inner-membrane left out.



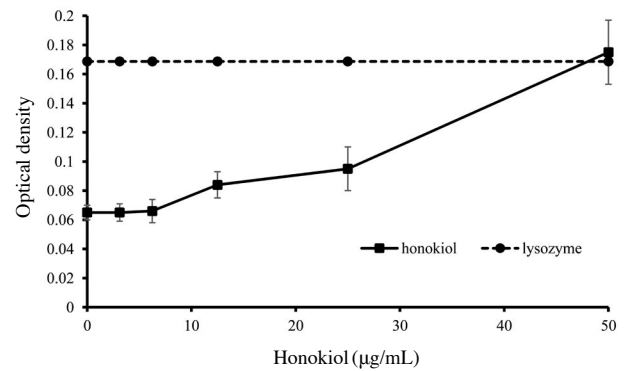
**Fig. 1.** Antimicrobial susceptibility test results based on agar diffusion analysis shown with image.

### Crystal violet assay

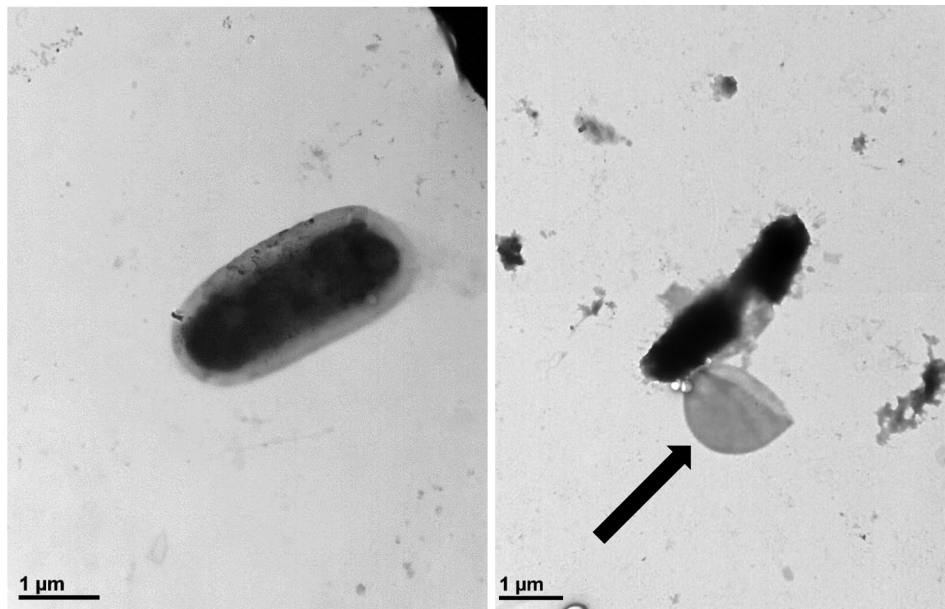
Since biofilm is so crucial to bacteria survival, biofilm disturbance of honokiol was examined using crystal violet assay (Fig. 4). There was no big difference between biofilm of *P. larvae* without honokiol and with 0.047  $\mu\text{g}/\text{mL}$  of honokiol. When the concentration was higher than 0.047  $\mu\text{g}/\text{mL}$ , biofilm formation kept decreasing as the concentration of honokiol increased. All of the results appeared in dose-dependent manner.

### Checkerboard assay

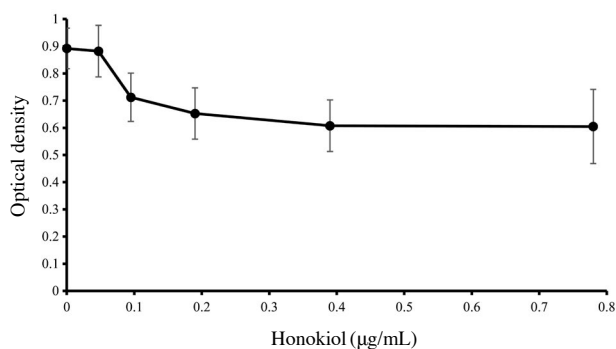
Synergy of honokiol with two antibiotics, terramycin



**Fig. 2.** Honokiol induced the leakage of 260 nm absorbing substances.



**Fig. 3.** TEM image of *P. larvae* after honokiol treatment. The left image is *P. larvae* without honokiol and the right image is *P. larvae* after honokiol treatment. The arrow indicates a membrane of *P. larvae*.



**Fig. 4.** Changes of biofilm formation after honokiol treatment.

**Table 4.** Synergy effect of honokiol with terramycin and tetracycline

Compound	MIC alone	MIC in combination	FIC index value
Honokiol	12.5 µg/mL	3.12 µg/mL	0.5
Terramycin	25 µg/mL	6.25 µg/mL	
Honokiol	12.5 µg/mL	3.12 µg/mL	0.5
Tetracycline	25 µg/mL	6.25 µg/mL	

and tetracycline, was determined using checkerboard assay (Table 4). When they were treated individually, the MIC value was 12.5 µg/mL, 25 µg/mL and 25 µg/mL respectively. However, when antibiotics were treated together with honokiol, MIC value has dropped to 3.12 µg/mL, 6.25 µg/mL and 6.25 µg/mL respectively. Both checkerboard assay of honokiol & terramycin and honokiol & tetracycline had FIC value of 0.5. According to previous studies (Schwalbe, 2007), It can be considered as additive compound of two antibiotics.

## DISCUSSION

The use of honokiol has been studied all around east Asia where *M. obovata* can be easily found. It is considered as a medicine for stomachache for a long period of time (Huang, 1998). Recently, its variety of effects has been discovered. Its anxiolytic activity was revealed (Maruyama *et al.*, 1998), and its antiangiogenic and antitumor activity has also reviewed (Fried and Arbiser, 2009). In this study, we have confirmed that honokiol can be an effective antibacterial agent against *P. larvae*.

Our results established that honokiol control the growth of *P. larvae* effectively. It is worth to note that

honokiol had greater antibacterial activity against *P. larvae* than terramycin and tetracycline with half of MIC value of those two antibiotics. To use honokiol as a chemical to control the growth of *P. larvae*, we might expect it is used as liquid solution and sprayed directly to the bacteria. When honokiol was treated with terramycin and tetracycline, the use of chemical was lessened. Like explained above, honokiol could be great antibacterial compound because it may slow or even stop the rise of *P. larvae* with antibiotics resistance.

Our results also demonstrated the target of honokiol against *P. larvae*. We showed that honokiol ruptures the membrane, resulting the bursting of the cell and it could be the reason why honokiol had low MIC value. From the images of TEM, it was curious how honokiol was so effective even it was treated only for an hour. It greatly induced the bursting of cells fast, proving its time-effective antimicrobial activity. From the studies done before, we could predict the ability of honokiol to rupture the membrane of *P. larvae* comes from its structure (Woodbury *et al.*, 2013). Since honokiol is classified as polyphenol family, it could easily merge with cell membrane proteins by interactions between molecules like hydrogen bonding or hydrophobic interactions. Identifying the exact mechanism of honokiol interacting membrane needs to be identified in further study. Its ability to interrupt biofilm formation were surprisingly effective considering the amount of honokiol used for the crystal violet assay. Considering that honokiol already had its ability interrupt biofilm formation at concentrations only over 0.047 µg/mL, it is now confirmed to be a considerably efficient anti-biofilm agent.

The safety of honokiol was not demonstrated in this study. In view of commercial use of this compound, its cytotoxicity will be measured and *in vivo* experiments using honeybees will be performed in further study.

## ACKNOWLEDGEMENTS

This research was supported by the IPET (116094-03-1-SB010).

## FOOTNOTES

Financial Disclosure: This research was conducted in the absence of any commercial or financial relation-

ships that could be construed as posing potential conflicts of interest.

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